

CHEMISTRY

(Minor)

(6th Semester)

Course No.: CHM-DSM-351

Practical

(Inorganic, Organic and Physical Chemistry)

Contact Hours: 60; Credits: 04

Full Marks = 100[End Semester Exam (70) Internal Assessment (30)]

Pass Marks = 40 [End Semester Exam (28) Internal Assessment(12)]

Examination Time: 12 hours (2 days)

Section-A (Inorganic Chemistry)

1. Qualitative Inorganic Analysis

25 Marks

Qualitative analysis of inorganic mixtures containing 2 anions and 2 cations without interfering radicals.

Section-B (Organic Chemistry)

2a. Systematic Qualitative Organic Analysis of Organic Compounds possessing monofunctional groups (-COOH, phenolic, aldehydic, ketonic, amide, nitro, amines). **15marks**

2b. Organic Preparation and purification:

10 marks

Organic preparation and reactions

- i) Nitration of acetanilide/ nitrobenzene/ salicylic acid
- ii) Bromination of phenol/aniline
- iii) Oxime/ 2,4-dinitrophenylhydrazone of aldehyde/ ketone.
- iv) Benzil from benzoin
- v) Benzilic acid from benzil
- vi) Benzoylation of Phenol/ aniline
- vii) Iodoform from acetone

viii)

Purification of organic compounds

- i) Decolorization of crude sulphanilic acid (recrystallization using animal charcoal)
- ii) Recrystallization of benzoic acid from hot water/ ethanol.
- iii) Recrystallization of Acetanilide from boiling water
- iv) Purification of naphthalene/ camphor/phthalic acid (by sublimation)



Section-C (Physical Chemistry)

3. Any one experiment out of the following can set in examination 20Marks

- vi. To determine the surface tension of glycerol/acetic acid Solutions at different concentrations and construction of graph.
- vii. To determine the viscosity of glycerol/acetic acid Solutions at different concentrations and construction of graph.
- viii. pH-metric titration of strong acid vs strong base.
 - ix. Conductometric titration of strong acid vs strong base.
 - x. To determine the solubility of benzoic acid at different temperature and to determine ΔH of the dissolution process.

Internal Assessment

7. Viva-voce
8. Regularity in maintenance of Lab Note Book
9. Attendance
10 Marks

Reference Book

- i. Vogel, A. I., A Textbook of Quantitative Inorganic Analysis, ELBS. 1978
- Khosla, B. D.; Garg, V. C. & Gulati, A., Senior Practical Physical Chemistry, R. Chand & Co.: New Delhi (2011).
- iii. Nad, A.K., Mahapatra, B., Ghoshal, A., An Advanced Course in Practical Chemistry, New Central Book Agency (P) Ltd., Kolkata, India.
- iv. Ahluwalia, V. K. & Aggarwal, R. Comprehensive Practical Organic Chemistry, Universities Press.
