



**2023/TDC(CBCS)/EVEN/SEM/
CHMHCC-401T/334**

TDC (CBCS) Even Semester Exam., 2023

CHEMISTRY

(Honours)

(4th Semester)

Course No. : CHMHCC-401T

(Coordination Chemistry and its Application)

Full Marks : 50

Pass Marks : 20

Time : 3 hours

*The figures in the margin indicate full marks
for the questions*

SECTION—A

Answer any *ten* of the following questions : $2 \times 10 = 20$

1. What is effective atomic number (EAN) of a complex? Find out the EAN of cobalt in $[\text{Co}(\text{NH}_3)_6]^{3+}$. 1+1=2
2. What is inner-orbital complex? Show hybridization of inner-orbital complex taking a suitable example. 1+1=2



(2)

3. Sketch tetragonal distortion of NiF_6^{4-}
(a d^8 arrangement in weak octahedral field).
4. (a) Write the formula of the compound sodium iron (III) hexacyanoferrate (II).
(b) Write the IUPAC name of the following compound :
 $[\text{Cr}(\text{en})_3][\text{Co}(\text{CN})_6]$ 1+1=2
5. $[\text{Co}(\text{NH}_3)_5\text{NO}_2]\text{Cl}_2$ can be prepared in two isomeric forms. Show the structures and comment on the type of isomerism.
6. What is an ambidentate ligand? How does it differ from polydentate ligand? 1+1=2
7. Cu and Au should be included in transition series, although having d^{10} configuration. Give arguments.
8. Give the reason for the different magnetic behaviours for the first-row transition elements compared to second- and third-row transition elements.

(3)

9. "The atomic radii of Zr and Hf are almost same." Explain.
10. Which lanthanide elute first in ion-exchange method? Give reason.
11. Why is Ce^{3+} solution colourless but Ce^{4+} solution yellow in colour?
12. Give the products for the following reactions : 1+1=2
(i) $\text{EuCl}_3 + \text{H}_2 \xrightarrow{\text{Zn/Hg}} [\text{A}]$
(ii) $\text{Ce}(\text{OH})_3 + \frac{1}{2}\text{O}_2 \xrightarrow{\Delta} [\text{B}]$
13. Is haemocyanin a non-haemeprotein? Write the metal ion, oxidation state and magnetic properties in haemocyanin.
14. Comment on the size and magnetism of Fe^{2+} in oxyhaemoglobin and deoxyhaemoglobin.
15. Write health disorders/diseases (human) originated from deficiency and excess of calcium ions.



(6)

23. (a) Why lanthanides show different magnetic properties than transition (*d*-block) elements? Give the equation for magnetic moment calculation of lanthanides. 2+1=3
- (b) Eu^{3+} and Sm^{3+} show anomalous magnetic behaviour than the other lanthanides. Explain. 3
24. (a) Describe the biological role of Zn in human body. 3
- (b) What is catalase? Write its function. How does it differ from peroxidase? 3
25. (a) How lead (Pb) toxicity happens? Write the effect of Pb poisoning on human health. Name the therapy by which Pb poisoning can be prevented. 1+2+1=4
- (b) What do you mean by essential and trace elements? Give examples. 2
