2023/TDC(CBCS)/ODD/SEM/ CHMHCC-302T/262

TDC (CBCS) Odd Semester Exam., 2023

CHEMISTRY

(Honours)

(3rd Semester)

Course No.: CHMHCC-302T

(Organic Chemistry)

Full Marks: 50 Pass Marks: 20

Time: 3 hours

The figures in the margin indicate full marks for the questions

SECTION-A

Answer ten questions, taking any two from each
Unit: 2×10=20

UNIT-I

1. Allyl chloride generally obeys S_N1 reaction during nucleophilic substitution reaction. Explain with proper reason.

(Turn Over)

2. Write the product of the reaction given below and justify it:

$$CH_3$$
— CH — CH — CH_3 $\xrightarrow{NaOH (aq)}$?

Explain why racemization takes place in S_N1 reaction.

UNIT-II

Write the products of the following reactions (any two):

4.
$$CH_3$$
— $CH=C-CH_3$ $1) O_3$ $? + ?$

5.
$$CH_3$$
— CH — CH_2 — CH_3 $\xrightarrow{conc. H_2SO_4}$? +?

6.
$$CH_3-CH_2-OH$$

$$\begin{array}{c}
\text{conc. } H_2SO_4 \\
180 °C
\end{array}?$$

$$\begin{array}{c}
140 °C \\
\text{conc. } H_2SO_4
\end{array}?$$

(Continued)

24J/261

UNIT-III

7. Identify A and B:

Ph—CHO + CH₃—CHO $\xrightarrow{\text{NaOH (aq)}} A$ $\xrightarrow{\Delta} B$

8. Identify C and D: Ph—CHO + HCN $\xrightarrow{11 \text{ NaOH}} C \xrightarrow{\text{H}^+/\text{H}_2\text{O}} D$

9. Which one of the following compounds will react with I_2 in presence of NaOH? Also write the product of the reactions.

UNIT-IV

- 10. What happens when maloic acid is treated with alkaline permanganate?
- 11. What happens when oxalic acid is heated with conc. H₂SO₄?
- 12. Identify A and B: CH_3 — CH_2 —COOH $\xrightarrow{Br_2}$ A $\xrightarrow{1) NaOH}$ $\xrightarrow{2) HCl}$ \xrightarrow{B} B

24J/261

(Turn Over)

UNIT-V

- Describe briefly the nature of the bond between carbon and magnesium in Grignard reagent.
- 14. What happens when ethane thiol is treated with acetic acid in presence of conc. H₂SO₄?
- 15. What happens when propyne is treated first with CH₃MgBr and followed by ethyl bromide?

SECTION-B

Answer five questions, taking one from each Unit:

6×5=30

UNIT-I

- 16. (a) Benzyl bromide is more reactive than cyclohexylmethyl bromide towards NaOH (aq) under the conditions of S_N1 process. Provide explanation.
 - (b) With the help of suitable example, write the mechanism and the stereochemistry of S_N2 reactions.

24J**/261**

(Turn Over)

2

2

2

2

(Continued)

3

- 17. (a) How the rate expression for $S_N 1$ and $S_N 2$ reactions differ?
 - (b) With suitable example, describe the mechanism of S_Ni reaction.
 - (c) CH₃—CH=CH—Cl is very unreactive towards nucleophile. Explain.

UNIT-II

- 18. (a) What happens when diethyl ether is treated with BF₃?
 - (b) How will you prepare picric acid from phenol? 2
 - (c) What happens when glycol is treated with acidic potassium permanganate? 2
- 19. Write the product and the mechanism of the reactions given below: 3+3=6

(i) OH
$$1) CO_2$$
 ?

24J/261

UNIT-III

20. Complete the following reactions and also write the mechanisms:
3×2=6

- 21. Explain the following reactions with mechanism:
 - (a) Baeyer-Villiger oxidation
 - (b) Perkin reaction
 - (c) Clemmensen reduction

UNIT-IV

- 22. Describe the following reactions with mechanism:
 - (a) Reformatsky reaction
 - (b) Hofmann bromamide degradation

24J/261 (Continued)

23. Complete the following reactions:

1×6=6

(i) HCOOH Fehling solution ?

(ii)
$$CH_3CH_2COOH$$
 $\xrightarrow{1) NH_3, \Delta}$?

(v) HO—CH—COOH
HO—CH—COOH
$$\xrightarrow{\text{excess of HI}} \Delta$$

UNIT-V

24. Starting from ethyl acetoacetate, obtain-

(a) 3-methylpentan-2-one;

(b) propanoic acid.

3×2=6

24J/261

(Turn Over)

- 25. Starting from diethyl malonate, obtain-
 - (a) crotonic acid;
 - (b) cyclopropane.

 $3 \times 2 = 6$

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