2023/TDC(CBCS)/ODD/SEM/ BTCHCC-303T/276

4. What is Slater's cule?

TDC (CBCS) Odd Semester Exam., 2023

BIOTECHNOLOGY

(Honours)

6. Write a note of retemps brise der Walla

Course No.: BTCHCC-303T

(Chemistry—I)

Full Marks: 50
Pass Marks: 20

Time: 3 hours

The figures in the margin indicate full marks for the questions

SECTION—A acted W SI

Answer ten questions, selecting any two from each
Unit: 2×10=20

UNIT-I

- 1. What is Pauli's exclusion principle?
- 2. Deduce de Broglie equation.
- 3. What is Aufbau's principle?

24J**/275**

(Turn Over)

TDC (CSCS) Odd Semester Exam.. 2623

- 4. What is Slater's rule?
- 5. What do you mean by effective nuclear charge?
- 6. Write a note on atomic radii (van der Waals).

Course No III—TINU

- 7. What is coefficient of viscosity?
- 8. What is the vapour pressure of a liquid?
- 9. Write down the physical properties of liquids.

UNIT-IV

- 10. What is viscosity of a liquid?
- 11. What are detergents?
- 12. What happens to the viscosity of water when a detergent is mixed to it?

Unit—V

13. What is an acid-base indicator?

14. What do you mean by strong and weak electrolytes?

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15. What is common-ion effect?

SECTION-B

21. What is surface tension? How is coefficient

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Answer five questions, selecting one from each
Unit: 6×5=30

UNIT-I

- 16. Explain Bohr's theory. What is Heisenberg's uncertainty principle? 4+2=6
- 17. What is Hund's rule of maximum multiplicity? Discuss Schrödinger's wave equation. 3+3=6

II—TINU 25. Explain the working of phenolphib

- 18. What do you mean by electronegativity?
 Write a note on shielding effect. 3+3=6
- 19. What do you mean by s-, p-, d- and f-block elements? Give examples of s-block and p-block elements.
 4+2=6

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(Turn Over)

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- 20. Discuss the effect of addition of solutes on surface tension.
- 21. What is surface tension? How is coefficient of viscosity determined? 2+4=6

UNIT-IV

- 22. Discuss the structure of water. Write briefly on cleansing action of detergents. 3+3=6
- 23. Explain why viscosity of liquid and gases varies with temperature.

Unit--V

- 24. What are the factors affecting degree of ionization? Add a note on ionization constant.

 3+3=6
- 25. Explain the working of phenolphthalein as an acid-base indicator with appropriate structure. What do you mean by weak acids and bases?

 4+2=6

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